COUNCIL DIRECTIVE

of 4 May 1976

amending for the second time Directive 65/66/EEC laying down specific criteria of purity for the preservatives authorized for use in foodstuffs intended for human consumption

(76/463/EEC)

THE COUNCIL OF THE EUROPEAN COMMUNITIES,

Having regard to the Treaty establishing the European Economic Community,

Having regard to Council Directive 64/54/EEC of 5 November 1963 on the approximation of the laws of the Member States concerning the preservatives authorized for use in foodstuffs intended for human consumption (¹), as last amended by Directive 76/462/EEC (²), and in particular Article 8 thereof,

Having regard to the proposal from the Commission,

Whereas Directive 65/66/EEC (³), as amended by Directive 67/428/EEC (⁴), laid down specific criteria of purity for the preservatives listed in the Annex to Directive 64/54/EEC; whereas that Annex was supplemented by Directive 71/160/EEC which added calcium sulphite to the list of authorized preservatives, by the Act of Accession which added methyl p-hydroxybenzoate and calcium bisulphite, by Directive 74/62/EEC which added formic acid, sodium formate, calcium formate and hexamethylenetetramine, by Directive 74/394/EEC which added thiabendazole and Directive 76/462/EEC which added the sodium derivative of methyl p-hydroxybenzoate, potassium nitrite and potassium propionate;

Whereas it is necessary to lay down specific criteria of purity for the eleven preservatives mentioned above,

HAS ADOPTED THIS DIRECTIVE :

Article 1

The Annex to Directive 65/66/EEC shall be amended as follows :

1. The following specifications shall be inserted between Nos E 217 and E 220:

E 218 p-hydroxybenzoate

White, almost odourless, crystalline powder
125 to 128 °C
Not less than 99.0 % expressed as $C_8H_8O_3$ after drying for two hours at 80 $^\circ\text{C}$
Not more than 0.05 %
Not more than 0.7 % expressed as p-hydroxybenzoic acid
Not more than 0.1 %
Not more than 0.5 % after drying for two hours at 80 $^{\circ}C$

E 219 Sodium derivative of methyl p-hydroxybenzoate

Appearance :	White hygroscopic powder
Melting range of methyl ester:	The white precipitate formed by acidifying with hydro
	chloric acid a 10 % (w/v) aqueous solution of the sodium

chloric acid a 10 % (w/v) aqueous solution of the sodium derivative of methyl p-hydroxybenzoate (using litmus paper as indicator) shall, when washed with water and dried at 80 °C for two hours, have a melting range of 125 to 128° C.

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⁽²⁾ See page 31 of this Official Journal.

^{(&}lt;sup>1</sup>) OJ No 12, 27. 1. 1964, p. 161/64.

⁽³⁾ OJ No 22, 9. 2. 1965, p. 373/65.

^{(&}lt;sup>4</sup>) OJ No 148, 11. 7. 1967, p. 10.

Content :	Not less than 99.5 % of C ₈ H ₇ O ₃ Na calculated on the dry matter		
Moisture :	Not more than 5.0 % (Karl-Fischer)		
Sulphated ash :	40.0 to 44.5 % calculated on the dry matter		
pH (0·1 % solution in carbon dioxi free water) :	ide- Not less than 9.7 and not more than 10.3		
Salicylic acid :	Not more than 0.1 %		
2. The following specifications shall	be inserted between Nos E 225 and E 230:		
E 226 Calcium sulphite			
Appearance :	White crystals or white crystalline powder		
Content :	Not less than 95 % of CaSO ₃ . $2H_2O$ and not less than 39 % of SO ₂		
Sulphates :	Not more than 0.1 %, expressed as SO4		
Chlorides :	Not more than 0.05 % expressed as Cl		
Iron :	Not more than 0.005 %		
Selenium :	Not more than 10 mg/kg of the SO ₂ content		
E 227 Calcium hydrogen sulphite			
Appearance :	Clear greenish-yellow aqueous solution having a distinct odour of sulphur dioxide		
Content :	6 to 8 % (w/v) of sulphur dioxide and 2.5 to 3.5 % (w/v) of calcium oxide corresponding to 10 to 14 % (w/v) of calcium bisulphite $[Ca(HSO_3)_2]$		
Iron :	Not more than 30 mg/kg		
Selenium :	Not more than 10 mg/kg of the SO ₂ content		
3. The following specifications shall	be inserted between Nos E 232 and E 250		
E 233 2-(4-thiazolyl) benzimidazole (thiabendazole)			
E 233 2-(4-thiazolyl) benzimidaz	ole (thiabendazole)		
<i>E 233 2-(4-thiazolyl) benzimidaz</i> Appearance :	<i>ole</i> (thiabendazole) White, or almost white, odourless powder		
<i>E 233 2-(4-thiazolyl) benzimidaz</i> Appearance : Melting range :	ole (thiabendazole) White, or almost white, odourless powder 296 to 303 °C		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content :	ole (thiabendazole) White, or almost white, odourless powder 296 to 303 °C 98 to 101 % C ₁₀ H ₇ N ₃ S calculated on the anhydrous product		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content : Sulphated ash :	ole (thiabendazole) White, or almost white, odourless powder 296 to 303 °C 98 to 101 % C ₁₀ H ₇ N ₃ S calculated on the anhydrous product Not more than 0.2 %		
<i>E 233 2-(4-thiazolyl) benzimidaz</i> Appearance : Melting range : Content : Sulphated ash : Moisture :	 ole (thiabendazole) White, or almost white, odourless powder 296 to 303 °C 98 to 101 % C₁₀H₇N₃S calculated on the anhydrous product Not more than 0.2 % Not more than 0.5 % (Karl-Fischer) 		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content : Sulphated ash : Moisture : UV Absorption (0.0005 % w/v in N HCl)	ole (thiabendazole) White, or almost white, odourless powder 296 to 303 °C 98 to 101 % $C_{10}H_7N_3S$ calculated on the anhydrous product Not more than 0.2 % Not more than 0.5 % (Karl-Fischer) 0.1 $E \frac{1 \%}{1 \text{ cm}}$ at 302 ± 2 nm = 1 230 approximately		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content : Sulphated ash : Moisture : UV Absorption (0.0005 % w/v in N HCl) :	ole (thiabendazole) White, or almost white, odourless powder 296 to 303 °C 98 to 101 % C ₁₀ H ₇ N ₃ S calculated on the anhydrous product Not more than 0.2 % Not more than 0.5 % (Karl-Fischer) 0.1 $E \frac{1 \%}{1 \text{ cm}}$ at 302 ± 2 nm = 1 230 approximately $E \frac{1 \%}{1 \text{ cm}}$ at 258 ± 2 nm = 200 approximately		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content : Sulphated ash : Moisture : UV Absorption (0.0005 % w/v in N HCl) :	ole (thiabendazole) White, or almost white, odourless powder 296 to 303 °C 98 to 101 % $C_{10}H_7N_3S$ calculated on the anhydrous product Not more than 0.2 % Not more than 0.5 % (Karl-Fischer) 0.1 $E \begin{array}{c} 1 \ \% \\ 1 \ cm}$ at 302 ± 2 nm = 1 230 approximately $E \begin{array}{c} 1 \ \% \\ 1 \ cm}$ at 258 ± 2 nm = 200 approximately $E \begin{array}{c} 1 \ \% \\ 1 \ cm}$ at 243 ± 2 nm = 620 approximately		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content : Sulphated ash : Moisture : UV Absorption (0.0005 % w/v in N HCl) :	ole (thiabendazole) White, or almost white, odourless powder 296 to 303 °C 98 to 101 % C ₁₀ H ₂ N ₃ S calculated on the anhydrous product Not more than 0.2 % Not more than 0.5 % (Karl-Fischer) 0.1 $E \frac{1 \%}{1 \text{ cm}}$ at 302 ± 2 nm = 1 230 approximately $E \frac{1 \%}{1 \text{ cm}}$ at 258 ± 2 nm = 200 approximately $E \frac{1 \%}{1 \text{ cm}}$ at 243 ± 2 nm = 620 approximately Ratio of $\frac{\text{absorption at } 241 \text{ to } 245 \text{ nm}}{\text{absorption at } 300 \text{ to } 304 \text{ nm}} = 0.47 \text{ to } 0.53$		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content : Sulphated ash : Moisture : UV Absorption (0.0005 % w/v in N HCl) :	ole (thiabendazole)White, or almost white, odourless powder 296 to 303 °C 98 to 101 % $C_{10}H_7N_3S$ calculated on the anhydrous product Not more than 0.2 % Not more than 0.5 % (Karl-Fischer)0.1 $E \begin{array}{c} 1 \ \% \\ 1 \ cm \end{array}$ at 302 ± 2 nm = 1 230 approximately E $\begin{array}{c} 1 \ \% \\ 1 \ cm \end{array}$ at 258 ± 2 nm = 200 approximately E $\begin{array}{c} 1 \ \% \\ 1 \ cm \end{array}$ at 243 ± 2 nm = 620 approximately Ratio of $\begin{array}{c} \frac{absorption at 241 \ to 245 \ nm \\ absorption at 300 \ to 304 \ nm \end{array}$ = 0.47 to 0.53 Ratio of $\begin{array}{c} \frac{absorption at 256 \ to 260 \ nm \\ absorption at 300 \ to 304 \ nm \end{array}$		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content : Sulphated ash : Moisture : UV Absorption (0.0005 % w/v in N HCl) : Selenium :	<i>ole</i> (thiabendazole) White, or almost white, odourless powder 296 to 303 °C 98 to 101 % C ₁₀ H ₇ N ₃ S calculated on the anhydrous product Not more than 0.2 % Not more than 0.5 % (Karl-Fischer) 0.1 $E \frac{1 \%}{1 \text{ cm}}$ at 302 ± 2 nm = 1 230 approximately $E \frac{1 \%}{1 \text{ cm}}$ at 258 ± 2 nm = 200 approximately $E \frac{1 \%}{1 \text{ cm}}$ at 243 ± 2 nm = 620 approximately Ratio of absorption at 241 to 245 nm absorption at 300 to 304 nm = 0.47 to 0.53 Ratio of absorption at 256 to 260 nm Ratio of absorption at 300 to 304 nm		
E 233 2-(4-thiazolyl) benzimidaz Appearance : Melting range : Content : Sulphated ash : Moisture : UV Absorption (0.0005 % w/v in N HCl) : Selenium : E 236 Formic acid	<i>white, or almost white, odourless powder</i> 296 to 303 °C 98 to 101 % C ₁₀ H ₇ N ₃ S calculated on the anhydrous product Not more than 0.2 % Not more than 0.5 % (Karl-Fischer) 0.1 $E \frac{1 \%}{1 \text{ cm}}$ at 302 ± 2 nm = 1 230 approximately $E \frac{1 \%}{1 \text{ cm}}$ at 258 ± 2 nm = 200 approximately $E \frac{1 \%}{1 \text{ cm}}$ at 243 ± 2 nm = 620 approximately Ratio of $\frac{\text{absorption at } 241 \text{ to } 245 \text{ nm}}{\text{absorption at } 300 \text{ to } 304 \text{ nm}} = 0.47 \text{ to } 0.53$ Ratio of $\frac{\text{absorption at } 256 \text{ to } 260 \text{ nm}}{\text{absorption at } 300 \text{ to } 304 \text{ nm}} = 0.14 \text{ to } 0.18$ 10 mg/kg		

Content : Acetic acid : , colourless, highly corrosive liquid with a characterpungent odour less than 98.0 % of CH_2O_2 nore than 0.5 %

Sulphates :	Not more than 40 mg/kg, expressed as SO_4
Sulphites :	Dilute 25 ml of formic acid with 25 ml of water. Add 0.1 ml of 0.1 N iodine solution. The solution should retain a distinct yellow colour
Chlorides :	Not more than 50 mg/kg, expressed as Cl
Specific gravity :	1.216 to 1.220 (20/20 °C)
Non-volatile matter :	Not more than 0.05 %
Aldehydes :	A slightly alkaline 5 % solution, on heating must not give off a sharp or burnt smell
Formaldehyde :	Not more than 0.1 % of the formic acid content, deter- mined using chromotropic acid
Oxalic acid :	Not more than 0.5 % of the formic acid content deter- mined as calcium oxalate, expressed as oxalic acid
E 237 Sodium formate	
Appearance :	White crystalline powder
Content :	Not less than 98 % NaCHO ₂ after drying for two hours at 105 °C
Volatile matter :	Not more than 2 % by drying for two hours at 105 °C
Degree of acidity or alkalinity :	Neutralization of 1 g of sodium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH
Aldehydes :	A 5 % solution on heating must not give off a sharp or burnt smell
Formaldehyde :	Not more than 0.1 % of the sodium formate content, determined using chromotropic acid
Oxalic acid :	Not more than 0.5 % of the sodium formate content, determined as calcium oxalate, expressed as oxalic acid
E 238 Calcium formate	
<i>E 238 Calcium formate</i> Appearance :	White crystalline powder
<i>E 238 Calcium formate</i> Appearance : Content :	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C
<i>E 238 Calcium formate</i> Appearance : Content : Volatile matter :	White crystalline powder Not less than 98 % $CaC_2H_2O_4$ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C
<i>E 238 Calcium formate</i> Appearance : Content : Volatile matter : Water-insoluble matter :	White crystalline powder Not less than 98 % $CaC_2H_2O_4$ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 %
<i>E 238 Calcium formate</i> Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity :	White crystalline powder Not less than 98 % $CaC_2H_2O_4$ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH
<i>E 238 Calcium formate</i> Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes :	White crystalline powder Not less than 98 % $CaC_2H_2O_4$ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell
E 238 Calcium formate Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes : Formaldehyde :	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid
E 238 Calcium formate Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes : Formaldehyde : Oxalic acid :	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid Not more than 0.3 % of the calcium formate content, determined as calcium oxalate, expressed as oxalic acid
E 238 Calcium formate Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes : Formaldehyde : Oxalic acid :	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid Not more than 0.3 % of the calcium formate content, determined as calcium oxalate, expressed as oxalic acid
 <i>E 238 Calcium formate</i> Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes : Formaldehyde : Oxalic acid : <i>E 239 Hexamethylenetetramine</i> Appearance : 	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid Not more than 0.3 % of the calcium formate content, determined as calcium oxalate, expressed as oxalic acid
 <i>E 238 Calcium formate</i> Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes : Formaldehyde : Oxalic acid : <i>E 239 Hexamethylenetetramine</i> Appearance : Content : 	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid Not more than 0.3 % of the calcium formate content, determined as calcium oxalate, expressed as oxalic acid
E 238 Calcium formate Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Degree of acidity or alkalinity : Aldehydes : Formaldehyde : Oxalic acid : E 239 Hexamethylenetetramine Appearance : Content : Loss on drying :	 White crystalline powder Not less than 98 % CaC₂H₂O₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid Not more than 0.3 % of the calcium formate content, determined as calcium oxalate, expressed as oxalic acid Colourless, or white, crystalline powder Not less than 99 % C₆H₁₂N₄ Not more than 0.5 % after drying at 105 °C in vacuum over phosphorus pentoxide for two hours
 E 238 Calcium formate Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes : Formaldehyde : Oxalic acid : E 239 Hexamethylenetetramine Appearance : Content : Loss on drying : Sublimation point : 	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid Not more than 0.3 % of the calcium formate content, determined as calcium oxalate, expressed as oxalic acid Colourless, or white, crystalline powder Not less than 99 % C ₆ H ₁₂ N ₄ Not more than 0.5 % after drying at 105 °C in vacuum over phosphorus pentoxide for two hours Sublimes at about 260 °C
E 238 Calcium formate Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes : Formaldehyde : Oxalic acid : E 239 Hexamethylenetetramine Appearance : Content : Loss on drying : Sublimation point : Sulphated ash :	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid Not more than 0.3 % of the calcium formate content, determined as calcium oxalate, expressed as oxalic acid Colourless, or white, crystalline powder Not less than 99 % C ₆ H ₁₂ N ₄ Not more than 0.5 % after drying at 105 °C in vacuum over phosphorus pentoxide for two hours Sublimes at about 260 °C Not more than 0.05 %
E 238 Calcium formate Appearance : Content : Volatile matter : Water-insoluble matter : Degree of acidity or alkalinity : Aldehydes : Formaldehyde : Oxalic acid : E 239 Hexamethylenetetramine Appearance : Content : Loss on drying : Sublimation point : Sulphated ash : Sulphates :	White crystalline powder Not less than 98 % CaC ₂ H ₂ O ₄ after drying for two hours at 105 °C Not more than 2 % after drying for two hours at 105 °C Not more than 0.5 % Neutralization of 1 g of calcium formate in the presence of phenolphthalein must not require more than 0.5 ml of 0.1 N HCl or 0.1 N NaOH A 5 % solution on heating must not give off a sharp or burnt smell Not more than 0.1 % of the calcium formate content, determined using chromotropic acid Not more than 0.3 % of the calcium formate content, determined as calcium oxalate, expressed as oxalic acid Colourless, or white, crystalline powder Not less than 99 % C ₆ H ₁₂ N ₄ Not more than 0.5 % after drying at 105 °C in vacuum over phosphorus pentoxide for two hours Sublimes at about 260 °C Not more than 0.05 %, expressed as SO ₄

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E 249 Potassium nitrite

	Appearance :	White, or slightly yellow, deliquescent granules
	Content :	Not less than 95 % after drying for four hours over silica gel
4.	pH (5 % solution in carbon dioxide- free and ammonia-free water): The following specifications shall be ins	Not less than 6.0 and not more than 9.0 serted between Nos E 282 and E 290
	E 283 Potassium propionate	
	Appearance :	White crystalline powder
	Content :	Not less than 99 % after drying for two hours at 105 °C

Content :Not less than 99 % after drying for two hours at 105 °CVolatile substances :Not more than 4 % after drying for two hours at 105 °CWater-insoluble substances :Not more than 0.3 %Readily oxidizable substances :No trace

Not more than 30 mg/kg

Article.2

Member States shall make any amendments to their laws necessary to comply with this Directive within one year of its notification and shall forthwith inform the Commission thereof. The laws thus amended shall be brought into force not later than two years after such notification.

Article 3

This Directive is addressed to the Member States.

Done at Brussels, 4 May 1976.

Iron :

For the Council The President G. THORN